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<p>: 9 24 2018 <u>Медведев</u></p>	<p>: <u>Александр</u> /</p>
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45.03.03

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( -5; -5; -6; -9; -10; -15).

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	2. _____ ,  , , ,	-5	-
	3. _____   -	-6-	-

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	4. _____	-9	-
	5. _____  ( _____ ,  _____ ; _____ ,  _____ ;  _____ ,  _____ ,  _____ )	-10  ( _____ ,  _____ ; _____ ,  _____ ;  _____ ,  _____ ,  _____ )	-
	6. _____	-15	-
	1. _____	-5	-

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	11. _____ - ( _____ , _____ ; _____ , _____ ; _____ , _____ )	-10	,
	12. _____	-15	,
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	2. _____ ,	-5	,
	3. _____ -	-6	,
	4. _____	-9	,
	5. _____ ( , ; , , ; , , )	-10	,
	6. _____	-15	,
3-	7. . _____ ,	-5	,
	8. _____ ,	-5	,
	9. . _____		,

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<b>10.</b>	_____	-9	,
<b>11.</b>	_____  ( , ; , , ; , , )	-10	,
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**Comment on *the meanings of articles*:**

1. A chapter will be devoted to developing the concept of productivity that we are to employ in this study.
2. These criteria prove conclusively whether the compound is an unsymmetrical molecule or a mixture of two symmetrical compounds.
3. The marked steric ( ) effect appears to be due to extra methyl substituents in the side chain.

4. The extra stability due to conjugation ( ) is greater the more the  $\pi$ -electrons are shared in the system.
5. The quinoid ( ) structure requires that the two central phenyl groups be coplanar (=lying in one plane).
6. The reaction is most easily interpreted by imagining it to involve generation of hydroxyl radicals.
7. The impurities ( ), without question, should not exceed 2 per cent.
8. The two theories are not different in their essential content; only in details.
9. The two columns ([ ] ) are now supposed to operate at atmospheric outlet ( ) pressure.
10. Separation is complete after the two bands ( ) have moved apart a distance equal to their mean width.
11. The ability of the three double bonds to hydrogenate ( ) clearly argues against this view.
12. The procedure was dropped having given low yields of end products.
13. The synthesis of methylamine was a first objective of the research.
14. In contrast to the second reaction, a sixfold decrease in yield was observed.
15. The same result was obtained when the product was heated with a further 20 g. of ethyl acetate.
16. The use of dry gas is essential, since water is a substance that can not obviously be considered inert.
17. Ignition upon exposure to air is observed with quite a number of organometallic ( ) compounds.
18. There is every reason to conclude that the entropy ( ) of activation does not matter here.
19. Absorption of oxygen in titanium is sometimes initiated at a temperature as low as 300°.
20. Determination of constitution of the compound in question is not a simple matter.
21. Confirmation of the double bond types suspected must be sought elsewhere in the spectrum.
22. Removal of the hydroxyl group requires little activation.
23. It can be seen at a glance that toluene does not readily undergo autoxidation ( ).

8-16 , 50% :

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- [www.pnas.org/](http://www.pnas.org/) (Proceedings of the National Academy of Sciences of the United States of America),
  - <http://evolbiol.ru/paperlist.htm>,
  - <http://palaeos.com/>
- 5) Windows 8 Russian. Windows Professional 8 Russian Upgrade;  
6) Microsoft Office Standard 2013 Russian.

**6.** -

<i>1</i>	<i>2</i>	<i>3</i>
<p>1.</p> <p>: 4,</p> <p>9( .</p> <p>19, . . , 1)</p>		<p><b>4</b></p> <p>’ -</p> <p>’ - 12 .</p> <p>’</p> <p>( )</p> <p><b>9</b></p> <p>’ -</p>
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<p>: 13</p> <p>( )</p>		<p>Windows 8 Russian. Windows Professional 8 Russian Upgrade. 104</p> <p>17.06.2013 .</p> <p>Microsoft Office Standard 2013 Russian. 114 12.11.2014 .</p>

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2	<p><b>2:</b> -</p> <p>2.1 : -</p> <p>2.2 : -</p>		4		15	<p>) 1,2</p> <p>) 1,2</p>	<p>, ,</p> <p>) 1: 13,</p> <p>13.1</p> <p>2: 2, 3</p> <p>) 4: 1</p> <p>3: 6</p>	
	<p><b>3:</b> -</p> <p>-</p> <p>3.1 : -</p> <p>3.2 : -</p> <p>3.3 : -</p>		6		23, 8	<p>) 1,2</p> <p>) 3,</p>	<p>, ,</p> <p>) 2: 4</p> <p>) 1: 4</p> <p>2: IV,</p> <p>1.1 – 1.5, 3-7</p>	

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	4	4	0	16
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	18	1	0	18
<b>3</b>				36
				10
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